

X3.17a

PHYS-107 (17) (Kaldon-37944)
WMU - Fall 2004
Exam 3A - 100,000 points

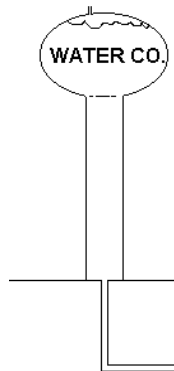
107

Name _____ SOLUTION _____

Book Title _____
Rev. 11/14/04 Su.A6b

State Any Assumptions You Need To Make – Show All Work – Circle Any Final Answers
Use Your Time Wisely – Work on What You Can – Be Sure to Write Down Equations
Short Answers Should Be Short! – Feel Free to Ask Any Questions

EXAM 3 [FORM - A]
PHYS-107 (KALDON-17)
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“Certified free of microbes, dust particles,
old tires, pigeon droppings and teenage
pranksters by the Department of Homeland
Security.”

*Your Friendly Neighborhood
Water Company...*

“Heat Waves” (50,000 points) **Multiple-Guess-Pick-The-Best-Answer-Fill-In-The-Bubbles**

1.) (a) A good can of beans hisses as air rushes in if the absolute pressure inside is _____ the outside.
A = Less Than B = Same As C = More Than D = None of these
The pressure inside the can must be lower. A B C D

(b) A putrid can of beans hisses as air rushes out if the gauge pressure (inside - outside) is _____ zero.
A = Less Than B = Same As C = More Than D = None of these
The pressure inside the can must be higher. A B C D

(c) A boat of mass 150. kg has dimensions 1.0 m × 0.50 m × 1.5 m. What is the maximum load that can be carefully placed in this boat before it sinks?
A = 150. kg B = 500. kg C = 600. kg D = 750. kg E = 1000. kg
F = None of these A B C D E F
V = 0.750m³, so m_{max} = 750.kg. Subtract 150.kg for boat, 600.kg for load.

(d) A 9” diameter pipe narrows to 1” diameter. Water flowing in the wide pipe is _____ as fast as the narrow pipe.
A = 1/3rd B = 1/9th C = just D = 3 times E = 9 times
F = None of these Use Continuity Equation. $9^2 = 81$. A B C D E F

In parts (e)-(g), select which of the laws and definitions of Thermodynamics is best used to solve or describe the problem.

(e) If you had an engine with a Second Law Efficiency equal to 100%, you would violate _____.
A = Thermo 1st law B = Thermo 2nd law C = Entropy
D = Actual Efficiency E = Carnot Efficiency F = None of these
Q_C cannot be zero, so 2nd Law Efficiency is never 100%. A B C D E F

(f) If you had an engine with a Second Law Efficiency equal to 110%, you would violate _____.
A = Thermo 1st law B = Thermo 2nd law C = Entropy
D = Actual Efficiency E = Carnot Efficiency F = None of these
At 110%, you're getting more Work than Q_H put in. A B C D E F

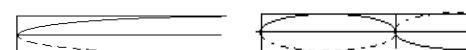
(g) Monitoring the temperature of the hot and cold reservoirs allows you to find the _____.
A = Thermo 1st law B = Thermo 2nd law C = Entropy
D = Actual Efficiency E = Carnot Efficiency F = None of these
Definition of Carnot Efficiency A B C D E F

Fred blows into a musical instrument and it sounds with a frequency $f = 500$. Hz. He blows harder and it sounds with a frequency $f = 1500$. Hz. The speed of sound in air is 343 m/s. Use $v = f \lambda$

(h) The wavelength of the 500. Hz tone is _____.
A = 0.229 m B = 0.343 m C = 0.500 m
D = 0.686 m E = 0.916 m F = None of these A B C D E F

(i) For the 1500. Hz tone, $\lambda =$ _____.
A = 0.229 m B = 0.343 m C = 0.500 m
D = 0.686 m E = 0.916 m F = None of these A B C D E F

(j) If one of these standing waves is the Fundamental and the other one is the 1st Overtone, then the pipe in this instrument is:
A = Closed at both ends B = Open at both ends
C = Closed at one end and open at the other end D = None of these A B C D
Diagrams for Problem 1(j): Fundamental 1st Overtone



One Million Gallons of Pure Clean Sweet Fresh Water (50,000 points)

2.) A city water tank contains 1,000,000 gallons of water (3.80×10^6 L volume, 3.80×10^6 kg mass) at a temperature of 9.00°C (48.2°F). (a) When a valve is opened at the bottom of the tank, water comes out at a speed of 20.0 m/s. What is the height h to the top of the water in the tank? The properties of water include: $c_{\text{water}} = 4186 \text{ J/kg}\cdot^\circ\text{C}$, $L_f = 334,000 \text{ J/kg}$, $L_v = 2,260,000 \text{ J/kg}$, $\rho_{\text{water}} = 1000 \text{ kg/m}^3$ and $\beta_{\text{water}} = 210 \times 10^{-6} \text{ }^\circ\text{C}^{-1}$.

$$P_1 + \rho g h_1 + \frac{1}{2} \rho v_1^2 = P_2 + \rho g h_2 + \frac{1}{2} \rho v_2^2$$

$$\rho g h_1 = \frac{1}{2} \rho v_2^2$$

$$h_1 = \frac{v_2^2}{2g} = \frac{(20.0 \text{ m/s})^2}{2(9.81 \text{ m/s}^2)} = 20.39 \text{ m}$$

(b) During the day, the temperature of the water rises from 9.00°C to 15.00°C (59.0°F). Find the new volume of the tank of water. For the purposes of this question only, give your answer to 5 significant figures.

$$\Delta V = \beta V_0 \Delta T$$

$$= (210 \times 10^{-6} \text{ }^\circ\text{C}^{-1})(3.80 \times 10^6 \text{ L})(15.00^\circ\text{C} - 9.0^\circ\text{C})$$

$$= +4788 \text{ L}$$

$$V = V_0 + \Delta V$$

$$= 3.80 \times 10^6 \text{ L} + 4788 \text{ L}$$

$$= 3,804,788 \text{ L} = 3,804,800 \text{ L}$$

Note that you can leave the volume in liters (L), because the only requirement is that V_0 and ΔV have the same units.

(c) How much heat energy Q did the water have to absorb during this expansion?

$$Q = mc\Delta T = (3.80 \times 10^6 \text{ kg})(4186 \text{ J/kg}\cdot^\circ\text{C})(+6.00^\circ\text{C})$$

$$= 95,440,000,000 \text{ J or } 9.544 \times 10^{10} \text{ J}$$

(d) During the night a pump is used to refill the tank, replacing the water used during the day. The 13.1 hp engine driving the pump does a useful work of 9780 J each second, while wasting 5470 J of heat energy each second. The Second Law Efficiency of this engine is 97.0% (0.970). Find the Actual Efficiency of this engine.

$$Q_H = W + Q_C$$

$$= 9780 \text{ J} + 5470 \text{ J} = 15,250 \text{ J}$$

$$\mathcal{E}_{\text{Actual}} = \frac{W}{Q_H} = \frac{9780 \text{ J}}{15,250 \text{ J}} = 0.6413 \text{ or } 64.13\%$$

In case you were wondering, $13.1 \text{ hp} = 9770 \text{ W} = 9770 \text{ J/s} \approx 9780 \text{ J/s}$.
(Dr. Phil worked the problem backwards.)

(e) Find the Carnot Efficiency of this engine. If you didn't get an answer to (d), use $\mathcal{E}_{\text{actual}} = 40.0\%$.

$$\mathcal{E}_{\text{2nd Law}} = \frac{\mathcal{E}_{\text{Actual}}}{\mathcal{E}_{\text{Carnot}}}$$

$$\mathcal{E}_{\text{Carnot}} = \frac{\mathcal{E}_{\text{Actual}}}{\mathcal{E}_{\text{2nd Law}}} = \frac{0.6413}{0.970} = 0.6611 \text{ or } 66.11\%$$

For those of you looking only for temperatures, you forgot we had two equations with $\mathcal{E}_{\text{Carnot}}$ in them.